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Introductory Chemistry-Nivaldo J. Tro 2014-01-01 See how chemistry is relevant to your life Now in its fifth edition, Introductory Chemistry continues to foster deep engagement in the course by showing how chemistry manifests in your daily life. Author Nivaldo Tro draws upon his classroom experience as an award-winning instructor to extend chemistry from the laboratory to your world, with relevant applications and a captivating writing style. Closely integrated with the fifth edition of Introductory Chemistry, MasteringChemistry® gives you the tools you need to succeed in this course. This program provides you a better learning experience. It will help you to:

- Personalize learning with MasteringChemistry®: This data-validated online homework, tutorial, and assessment program helps you quickly master concepts, and enables instructors to provide timely intervention when necessary.
- Achieve deep conceptual understanding: Several new Conceptual Checkpoints and Self-Assessment Quizzes help you better grasp key concepts.
- Develop problem-solving skills: A step-by-step framework encourages you to think logically rather than simply memorize formulas. Additional worked examples, enhanced with audio and video, reinforce challenging problems.
- Maintain interest in chemistry: The inclusion of concrete examples of key ideas throughout the program keeps you engaged in the material.

Note: If you are purchasing the standalone text or electronic version, MasteringChemistry does not come automatically packaged with the text. To purchase MasteringChemistry please visit: www.masteringchemistry.com or you can purchase a package of the physical text + MasteringChemistry by searching for 9780321910073 / 0321910079. MasteringChemistry is not a self-paced technology and should only be purchased when required by an instructor.

Microbiology (Questions and Answers), 5e-Purshotam Kaushik & Kirti Kaushik Microbiology is an engaging textbook presenting balanced and comprehensive account of major areas of microbiology in the form of questions and answers. This question-answer approach to present complex topics and theories of microbiology regarding cellular and non-cellular microorganisms, microbial genetics and molecular biology in higher plants and animals, makes the subject interesting and easily comprehensible for the students.

Introductory Chemistry: An Active Learning Approach-Mark S. Cracolice 2015-01-01 Teach the course your way with INTRODUCTORY CHEMISTRY, 6e. Available in multiple formats (standard paperbound edition, loose-leaf edition, digital MindTap Reader edition, and a hybrid edition, which includes OWLv2), this text allows you to tailor the order of chapters to accommodate your particular needs, not only by presenting topics so they never assume prior knowledge, but also by including any necessary preview or review information needed to learn that topic. The authors' question-and-answer presentation, which allows students to actively learn chemistry while studying an assignment, is reflected in three words of advice and encouragement that are repeated throughout the book: Learn It Now! This edition integrates new technological resources, coached problems in a two-column format, and enhanced art and photography, all of which dovetail with the authors' active learning approach. Even more flexibility is provided in the new MindTap Reader edition, an electronic version of the text that features interactivity, integrated media, additional self-test problems, and clickable key terms and answer buttons for worked examples. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Survival Guide to General Chemistry-Patrick E. McMahon 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed step-by-step methods and procedures for solving the major types of problems in general chemistry.

Explanations, instructional process sequences, solved examples and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid/base reactions which often persists through working with acid/base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

Dr. Salm's Chemistry Problem Drill Book-Robert J. Salm 2005-01 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College. This work will give you practice with the major problem types as you prepare for finals and standardized tests.

Multiple Solution Methods for Teaching Science in the Classroom-Stephen DeMeo 2008 For the first time in science education, the subject of multiple solution methods is explored in book form. While a multiple method teaching approach is utilized extensively in math education, there are very few journal articles and no texts written on this topic in science. Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons. First it challenges the practice by teachers that one specific method should be used when solving problems. Secondly, it calls into question the belief that multiple methods would confuse students and retard their learning. Using a case study approach and informed by research conducted by the author, this book claims that providing students with a choice of methods as well as requiring additional methods as a way to validate results can be beneficial to student learning. A close reading of the literature reveals that time spent on elucidating concepts rather than on algorithmic methodologies is a critical issue when trying to have students solve problems with understanding. It is argued that conceptual understanding can be enhanced through the use of multiple methods in an environment where students can compare, evaluate, and verbally discuss competing methodologies through the facilitation of the instructor. This book focuses on two very useful methods: proportional reasoning (PR) and dimensional analysis (DA). These two methods are important because they can be used to solve a large number of problems in all of the four academic sciences (biology, chemistry, physics, and earth science). This book concludes with a plan to integrate DA and PR into the academic science curriculum starting in late elementary school through to the introductory college level. A challenge is presented to teachers as well as to textbook writers who rely on the single-method paradigm to consider an alternative way to teach scientific problem solving.

Chemistry-Theodore L. Brown 1999-06-01

STOICHIOMETRY AND PROCESS CALCULATIONS-K. V. NARAYANAN 2006-01-01 This textbook is designed for undergraduate courses in chemical engineering and related disciplines such as biotechnology, polymer technology, petrochemical engineering, electrochemical engineering, environmental engineering, safety engineering and industrial chemistry. The chief objective of this text is to prepare students to make analysis of chemical processes through calculations and also to develop in them systematic problem-solving skills. The students are introduced not only to the application of law of combining proportions to chemical reactions (as the word 'stoichiometry' implies) but also to formulating and solving material and energy balances in processes with and without chemical reactions. The book presents the fundamentals of chemical engineering operations and processes in an accessible style to help the students gain a thorough understanding of chemical process calculations. It also covers in detail the background materials such as units and conversions, dimensional analysis and dimensionless groups, property estimation, P-V-T behaviour of fluids, vapour pressure and phase equilibrium relationships, humidity and saturation. With the help of examples, the book explains the construction and use of reference-substance plots, equilibrium diagrams, psychrometric charts, steam tables and enthalpy composition diagrams. It also elaborates on thermophysics and thermochemistry to acquaint the students with the thermodynamic principles of energy balance calculations. Key Features :

- SI units are used throughout the book.
- Presents a thorough introduction to basic chemical engineering principles.
- Provides many worked-out examples and exercise problems with answers.
- Objective type questions included at the end of the book serve as useful review material and also assist the students in preparing for competitive examinations such as GATE.

Sinners in the Hands of an Angry God-Jonathan Edwards 1922

Electron Paramagnetic Resonance of Transition Ions-A. Abragam 2012-06-28 A reissue of a classic Oxford text. The book is designed to provide a comprehensive introduction to the subject of electron paramagnetic resonance.

Peterson's MCAT Success 2005-Stefan Bosworth 2004-10 UNIQUE SELLING FEATURES AND BENEFITS- Thorough science review- 3 full-length sample MCAT exams- Hundreds of practice problems with detailed explanations- CD-Rom containing an additional full-length test and a quick-check diagnostic test- Sample medical school interview questions and a summary of what to expect once you are in medical school

Programmed Problem-solving for First-year Chemistry-Arnold B. Loebel 1983

Three Cognitive Skills in Chemistry and Their Application to Stoichiometry-Ardas Ozsogomonyan 1977

Stoichiometry and Thermodynamics of Metallurgical Processes-Y. K. Rao 1985-10-31 This textbook provides a thorough and comprehensive introduction to stoichiometry and thermodynamics with special emphasis on applications to metallurgical processes. The author's approach is to introduce students early on to the fundamentals of the physical chemistry and thermodynamics of metallurgical processes and then gradually expand the treatment into progressively more advanced areas. Topics covered include the laws of thermodynamics, material and energy balances, gasification and combustion of fuels, the iron blast furnace, direct reduction reactors, nonferrous smelters, fluidized-bed roasters, the theory of solutions, chemical equilibrium, electrochemistry. Also included are over 150 worked examples and 450 exercises, many with solutions. The examples and exercises range from straightforward tests of theory to complex analyses of real processes. Every chapter is provided with a full and up-to-date set of references.

An Introduction to Chemistry-Mark Bishop 2002 Bishop's text shows students how to break the material of preparatory chemistry down and master it. The system of objectives tells the students exactly what they must learn in each chapter and where to find it.

Chemistry-James E. Brady 2009 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon, as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon-buckyballs, graphite, and diamond-are illustrated at the left, as is the molecule methane, CH₄, from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane, found in natural gas, to the complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement.

SourceBook-Mary Virginia Orna 1994 Contains thirty-six content modules covering a broad range of chemistry curriculum topics. Includes ideas, demonstrations, laboratory activities, teaching techniques and strategies for beginning and experienced teachers.

Computers in Education-Olivier Lecarme 1975

Teaching Science for Understanding-James J. Gallagher 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities.

Kelly/Grimes IBM PC Compatible Computer Directory-Brian W. Kelly 1985

Basic Chemical Principles-Edward I. Peters 1988-01-01

Practice Makes Perfect Chemistry-Marian DeWane 2011-05-10 Don't be confused by chemistry. Master this science with practice, practice, practice! Practice Makes Perfect: chemistry is a comprehensive guide and workbook that covers all the basics of chemistry that you need to understand this subject. Each chapter focuses on one major topic, with thorough explanations and many illustrative examples, so you can learn at your own pace and really absorb the information. You get to apply your knowledge and practice what you've learned through a variety of exercises, with an answer key for instant feedback. Offering a winning formula for getting a handle on science right away, Practice Makes Perfect: chemistry is your ultimate resource for building a solid understanding of chemistry fundamentals.

Practice Makes Perfect Chemistry Review and Workbook, Second Edition-Marian DeWane 2018-12-28 The Winning Equation for Success in Chemistry is Practice, Practice, Practice! This book will help you apply concepts and see how chemistry topics are interconnected. Inside are numerous lessons to help you better understand the subject. These lessons are accompanied by dozens of exercises to practice what you've learned, along with a complete answer key to check your work. Throughout this book you will learn the terms to help you understand chemistry, and you will expand your knowledge of the subject through hundreds of sample questions and their solutions. With the lessons in this book, you will find it easier than ever to grasp chemistry concepts. And with a variety of exercises for practice, you will gain confidence using your growing chemistry skills in your classwork and on exams. YOU'LL BE ON YOUR WAY TO MASTERING THESE TOPICS AND MORE

- Atomic structure
- The periodic table
- Chemical formulas
- Chemical reactions
- Mass and mole relationships
- Gas laws
- Solutions
- Acids and bases
- Thermochemistry

A brand-new chapter on the structure of molecules

Chemical Problem-solving by Dimensional Analysis-Arnold B. Loebel 1978

Elements of General Chemistry-Jay A. Young 1960

Stoichiometry-B. I. Bhatt 1996

Numerical Chemistry-

Instructor's manual and solutions to problems to accompany General chemistry: principles and structures-James E. Brady 1975-04

Chemistry with Inorganic Qualitative Analysis-Therald Moeller 1980

Chemistry-John Christian Bailar 1984

A Stoichiometry Unit-David Callaghan 2004

How to Solve Word Problems in Chemistry-David E Goldberg 2001-06-26 In addition to having to master a vast number of difficult concepts and lab procedures, high school chemistry students must also learn, with little or no coaching from their teachers, how to solve tough word problems. Picking up where standard chemistry texts leave off, How to Solve Word Problems in Chemistry takes the fear and frustration out of chemistry word problems by providing students with easy-to-follow procedures for solving problems in everything from radioactive half-life to oxidation-reduction reactions.

Improving Student Comprehension of Stoichiometric Concepts-Connie Lynn Bannick Kemner 2007

Chemistry for Students and Parents-Roy Richard Sawyer 2013-12-15 If you are a parent struggling to help your child with chemistry homework, this is a short book that will help you. It covers key chemistry topics: Oxides, Bases, Acids, Salts, Equivalent proportions, Acid Base reactions, Weight and Volume problems, Equilibrium, Le Chatelier's Principle, Freezing and Boiling points, Balance Redox Reactions (30 examples with explanations), Stoichiometry (30 problems with answers and solutions). If you are student, read this book and you will prove to yourself that you can understand chemistry!

Chemistry-John McMurry 1998 A full-year course taken primarily by chemistry majors, other science majors (especially biology and pre-health), and engineering students. First introduced in 1995, McMurry/Fay's Chemistry is now recognized as one of the leading books in science education. The second edition refines the qualities that led to the text's success in the first place. "The text is a beautifully presented and well written general chemistry text. The chapter on gas laws, where the combination of narrative and illustrations leads the students to almost derive the Kinetic-Molecular theory on their own." (Mildred Hall, Clark State Community College.)

Chemistry-Steven S. Zumdahl 2003-07 Contains chapter discussions, key-term definitions, and practice chapter tests.

Introductory Chemistry-Steven S. Zumdahl 1999-07

Chemistry, Problem-Solving Worktext-James E. Brady 2002-06-13 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical substances and the changes, both chemical and physical, that they undergo. Fireworks: One of the ingredients is phosphorus, a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms. The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light. Carbon: The element is found in several forms, including the familiar diamond and another, recently discovered, sooty substance that consists of soccer-ball shaped molecules, often referred to as "buckyballs." Diamond is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth. Acetylene torch: Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen, a reaction between molecules of C₂H₂ and O₂ to give CO₂ and H₂O. Hot air balloon: The air that helps it rise is heated by the combustion of molecules of propane, each composed of three carbon and eight hydrogen atoms. Stormy weather: The evaporation of water serves to store energy provided by the sun. Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet.

Heath Chemistry Learning Guide-Kenneth Janda 2007-03-30

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