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Molecular Thermodynamics of Electrolyte Solutions-Lloyd L. Lee 2008-07-08 The introductory textbook provides an update on electrolyte thermodynamics with a molecular perspective. It is eminently suited as an introduction to the solution thermodynamics of ionic mixtures at the undergraduate and graduate level. It is also invaluable for the understanding and design in the engineering of natural gas treating and adsorption refrigeration with electrolytes.

Molecular Thermodynamics-Donald A. McQuarrie 1999-02-24 Evolved from McQuarrie and Simon's best-selling textbook, Physical Chemistry: A Molecular Approach, this text focuses in on the thermodynamics portion of the course. Although many of the chapters in Molecular Thermodynamics are similar to chapters in the larger physical chemistry text, new material has been added throughout along with three entirely new chapters on "The Energy Levels of Atoms and Molecules," "Thermodynamics of Electrochemical Cells," and "Nonequilibrium Thermodynamics." The text also includes five short "Math Chapters," each with a special set of problems that will help students review and summarize the mathematical tools required to master the material. Worked examples and chapter-ending problems with solutions are also included throughout the book.

Molecular Thermodynamics of Fluid-Phase Equilibria-John M. Prausnitz 1998-10-22 The classic guide to mixtures, completely updated with new models, theories, examples, and data. Efficient separation operations and many other chemical processes depend upon a thorough understanding of the properties of gaseous and liquid mixtures.

Molecular Thermodynamics of Fluid-Phase Equilibria, Third Edition is a systematic, practical guide to interpreting, correlating, and predicting thermodynamic properties used in mixture-related phase-equilibrium calculations. Completely updated, this edition reflects the growing maturity of techniques grounded in applied statistical thermodynamics and molecular simulation, while relying on classical thermodynamics, molecular physics, and physical chemistry wherever these fields offer superior solutions. Detailed new coverage includes: Techniques for improving separation processes and making them more environmentally friendly. Theoretical concepts enabling the description and interpretation of solution properties. New models, notably the lattice-fluid and statistical associated-fluid theories. Polymer solutions, including gas-polymer equilibria, polymer blends, membranes, and gels. Electrolyte solutions, including semi-empirical models for solutions containing salts or volatile electrolytes. Coverage also includes: fundamentals of classical thermodynamics of phase equilibria; thermodynamic properties from volumetric data; intermolecular forces; fugacities in gas and liquid mixtures; solubilities of gases and solids in liquids; high-pressure phase equilibria; virial coefficients for quantum gases; and much more. Throughout, Molecular Thermodynamics of Fluid-Phase Equilibria strikes a perfect balance between empirical techniques and theory, and is replete with useful examples and experimental data. More than ever, it is the essential resource for engineers, chemists, and other professionals working with mixtures and related processes.

Introduction to Molecular Thermodynamics-Robert M. Hanson 2008-07-21 Starting with just a few basic principles of probability and the distribution of energy, Introduction to Molecular Thermodynamics takes students on an adventure into the inner workings of the molecular world like no other, from probability to Gibbs energy and beyond, following a logical step-by-step progression of ideas.

Molecular Theory of Solutions-Arieh Ben-Naim 2006-07-27 This book presents new and updated developments in the molecular theory of mixtures and solutions. It is based on the theory of Kirkwood and Buff which was published more than fifty years ago. This theory has been dormant for almost two decades. It has recently become a very powerful and general tool to analyze, study and understand any type of mixtures from the molecular, or the microscopic point of view. The traditional approach to mixture has been, for many years, based on the study of excess thermodynamic quantities. This provides a kind of global information on the system. The new approach provides information on the local properties of the same system. Thus, the new approach supplements and enriches our information on mixtures and solutions.

Molecular Engineering Thermodynamics-Juan J. de Pablo 2014-07-10 Building up gradually from first principles, this unique introduction to modern thermodynamics integrates classical, statistical and molecular approaches and is especially designed to support students studying chemical and biochemical engineering. In addition to covering traditional problems in engineering thermodynamics in the context of biology and materials chemistry, students are also introduced to the thermodynamics of DNA, proteins, polymers and surfaces. It includes over 80 detailed worked examples, covering a broad range of scenarios such as fuel cell efficiency, DNA/protein binding, semiconductor manufacturing and polymer foaming, emphasizing the practical real-world applications of thermodynamic principles; more than 300 carefully tailored homework problems, designed to stretch and extend students' understanding of key topics, accompanied by an online solution manual for instructors; and all the necessary mathematical background, plus resources summarizing commonly used symbols, useful equations of state, microscopic balances for open systems, and links to useful online tools and datasets.

Molecular Thermodynamics of Nonideal Fluids-Lloyd L. Lee 2016-02-06 Molecular Thermodynamics of Nonideal Fluids serves as an introductory presentation for engineers to the concepts and principles behind and the advances in molecular thermodynamics of nonideal fluids. The book covers related topics such as the laws of thermodynamics; entropy; its ensembles; the different properties of the ideal gas; and the structure of liquids. Also covered in the book are topics such as integral equation theories; theories for polar fluids; solution thermodynamics; and molecular dynamics. The text is recommended for engineers who would like to be familiarized with the concepts of molecular thermodynamics in their field, as well as physicists who would like to teach engineers the importance of molecular thermodynamics in the field of engineering.

Molecular Thermodynamics of Copolymer Solutions and Blends-Toshiaki Hino 1995

Molecular Thermodynamics of Complex Systems-Xiaohua Lu 2008-12-21 With the development of science and technology, more and more complex materials such as porous materials, ion liquid, liquid crystals, thin films and colloids etc. are being developed in laboratories. However, it is difficult to prepare these advanced materials and use them on a large scale without some experience. Therefore, molecular thermodynamics, a method that laid emphasis on correlating and interpreting the thermodynamic properties of a variety of fluids in the past, has been recently employed to study the equilibrium properties of complex materials and establish thermodynamic models to analyze the evolution process of their components, structures and functions during the preparation process. In this volume, some important progress in this field, from fundamental aspects to practical applications, is reviewed. In the first chapter of this volume, Prof. Jianzhong Wu presents the application of Density Functional theory (DFT) for the study of the structure and thermodynamic properties of both bulk and inhomogeneous fluids. This chapter presents a tutorial overview of the basic concepts of DFT for classical systems, the mathematical relations linking the microstructure and correlation functions to measurable thermodynamic quantities, and the connections of DFT with conventional liquid-state theories. While for pedagogy the discussion is limited to one-component simple fluids, similar ideas and concepts are directly applicable to mixtures and polymeric systems of practical concern. This chapter also covers a few theoretical approaches to formulate the thermodynamic functional.

Molecular Thermodynamics of Proteins in Aqueous Solutions of Concentrated Electrolyte-Robin Andrew Curtis 2000

Solution Thermodynamics and Its Application to Aqueous Solutions-Yoshikata Koga 2017-03-28 Solution Thermodynamics and its Application to Aqueous Solutions: A Differential Approach, Second Edition introduces a differential approach to solution thermodynamics, applying it to the study of aqueous solutions. This valuable approach reveals the molecular processes in solutions in greater depth than that gained by spectroscopic and other methods. The book clarifies what a hydrophobe, or a hydrophile, and in turn, an amphiphile, does to H₂O. By applying the same methodology to ions that have been ranked by the Hofmeister series, the author shows that the kosmotropes are either hydrophobes or hydration centers, and that chaotropes are hydrophiles. This unique approach and important updates make the new edition a must-have reference for those active in solution chemistry. Unique differential approach to solution thermodynamics allows for experimental evaluation of the intermolecular interaction. Incorporates research findings from over 40 articles published since the previous edition. Numerical or graphical evaluation and direct experimental determination of third derivatives, enthalpic and volumetric AL-AL interactions and amphiphiles are new to this edition. Features new chapters on spectroscopic study in aqueous solutions as well as environmentally friendly and hostile water aqueous solutions.

Molecular Thermodynamics of Protein Interactions and Phase Behavior in Aqueous Electrolyte Solution-Christopher James Coen 1995

Molecular Thermodynamics of Fluid-phase Equilibria-J. M. Prausnitz 1969 97774-4 The classic guide to mixtures, completely updated with new models, theories, examples, and data. Efficient separation operations and many other chemical processes depend upon a thorough understanding of the properties of gaseous and liquid mixtures.

Molecular Thermodynamics of Fluid-Phase Equilibria, Third Edition is a systematic, practical guide to interpreting, correlating, and predicting thermodynamic properties used in mixture-related phase-equilibrium calculations. Completely updated, this edition reflects the growing maturity of techniques grounded in applied statistical thermodynamics and molecular simulation, while relying on classical thermodynamics, molecular physics, and physical chemistry wherever these fields offer superior solutions. Detailed new coverage includes: Techniques for improving separation processes and making them more environmentally friendly. Theoretical concepts enabling the description and interpretation of solution properties. New models, notably the lattice-fluid and statistical associated-fluid theories. Polymer solutions, including gas-polymer equilibria, polymer blends, membranes, and gels. Electrolyte solutions, including semi-empirical models for solutions containing salts or volatile electrolytes. Coverage also includes: fundamentals of classical thermodynamics of phase equilibria; thermodynamic properties from volumetric data; intermolecular forces; fugacities in gas and liquid mixtures; solubilities of gases and solids in liquids; high-pressure phase equilibria; virial coefficients for quantum gases; and much more. Throughout, Molecular Thermodynamics of Fluid-Phase Equilibria strikes a perfect balance between empirical techniques and theory, and is replete with useful examples and experimental data. More than ever, it is the essential resource for engineers, chemists, and other thermodynamic principles are discussed, along with predictive methods and molecular thermodynamics. This book is comprised of 12 chapters; the first of which introduces the reader to mathematical relationships, such as concentration variables, homogeneous functions, Euler's theorem, exact differentials, and method of least squares. The discussion then turns to partial molar quantities, ideal and nonideal solutions, and empirical expressions for predicting the thermodynamic properties of multicomponent mixtures from binary data. The chapters that follow explore binary and ternary mixtures containing only nonspecific interactions; the thermodynamic excess properties of liquid mixtures and ternary alcohol-hydrocarbon systems; and solubility behavior of nonelectrolytes. This book concludes with a chapter describing the use of gas-liquid chromatography in determining the activity coefficients of liquid mixtures and mixed virial coefficients of gaseous mixtures. This text is intended primarily for professional chemists and researchers, and is invaluable to students in chemistry or chemical engineering who have background in physical chemistry and classical thermodynamics.

Molecular thermodynamics-Richard Earl Dickerson 1969

Applied Chemical Engineering Thermodynamics-Dimitrios P. Tassios 2013-12-19 Applied Chemical Engineering Thermodynamics provides the undergraduate and graduate student of chemical engineering with the basic knowledge, the methodology and the references he needs to apply it in industrial practice. Thus, in addition to the classical topics of the laws of thermodynamics, pure component and mixture thermodynamic properties as well as phase and chemical equilibria the reader will find: - history of thermodynamics - energy conservation - intermolecular forces and molecular thermodynamics - cubic equations of state - statistical mechanics. A great number of calculated problems with solutions and an appendix with numerous tables of numbers of practical importance are extremely helpful for applied calculations. The computer programs on the included disk help the student to become familiar with the typical methods used in industry for volumetric and vapor-liquid equilibria calculations.

Problems and Solutions on Thermodynamics and Statistical Mechanics-Yung-kuo Lim 1990 Volume 5.

Journal of the American Chemical Society-American Chemical Society 1910 Proceedings of the Society are included in v. 1-59, 1879-1937.

Statistical Mechanics of Liquids and Solutions-Roland Kjellander 2019-07-30 The statistical mechanical theory of liquids and solutions is a fundamental area of physical sciences with important implications for many industrial applications. This book shows how you can start from basic laws for the interactions and motions of microscopic particles and calculate how macroscopic systems of these particles behave, thereby explaining properties of matter at the scale that we perceive. Using this microscopic, molecular approach, the text emphasizes clarity of physical explanations for phenomena and mechanisms relevant to fluids, addressing the structure and behavior of liquids and solutions under various conditions. A notable feature is the author's treatment of forces between particles that include nanoparticles, macroparticles, and surfaces. The book also provides an expanded, in-depth treatment of polar liquids and electrolytes.

Fluctuation Theory of Solutions-Paul E. Smith 2016-04-19 There are essentially two theories of solutions that can be considered exact: the McMillan-Mayer theory and Fluctuation Solution Theory (FST). The first is mostly limited to solutes at low concentrations, while FST has no such issue. It is an exact theory that can be applied to any stable solution regardless of the number of components and their concentrations, and the types of molecules and their sizes. Fluctuation Theory of Solutions: Applications in Chemistry, Chemical Engineering, and Biophysics outlines the general concepts and theoretical basis of FST and provides a range of applications described by experts in chemistry, chemical engineering, and biophysics. The book, which begins with a historical perspective and an introductory chapter, includes a basic derivation for more casual readers. It is then devoted to providing new and very recent applications of FST. The first application chapters focus on simple model, binary, and ternary systems, using FST to explain their thermodynamic properties and the concept of preferential solvation. Later chapters illustrate the use of FST to develop more accurate potential functions for simulation, describe new approaches to elucidate microheterogeneities in solutions, and present an overview of solvation in new and model systems, including those under critical conditions. Expert contributors also discuss the use of FST to model solute solubility in a variety of systems. The final chapters present a series of biological applications that illustrate the use of FST to study cosolvent effects on proteins and their implications for protein folding. With the application of FST to study biological systems now well established, and given the continuing developments in computer hardware and software increasing the range of potential applications, FST provides a rigorous and useful approach for understanding a wide array of solution properties. This book outlines those approaches, and their advantages, across a range of disciplines, elucidating this robust, practical theory.

A Textbook of Physical Chemistry-Arthur Adamson 2012-12-02 A Textbook of Physical Chemistry, Second Edition serves as an introductory text to physical chemistry. Topics covered range from wave mechanics and chemical bonding to molecular spectroscopy and photochemistry; ideal and nonideal gases; the three laws of thermodynamics; thermochemistry; and solutions of nonelectrolytes. The kinetics of gas-phase reactions; colloids and macromolecules; and nuclear chemistry and radiochemistry are also discussed. This edition is comprised of 22 chapters; the first of which introduces the reader to the behavior of ideal and nonideal gases, with particular emphasis on the van der Waals equation. The discussion then turns to the kinetic molecular theory of gases and the application of the Boltzmann principle to the treatment of molar polarization; dipole and magnetic moments; the phenomenology of light absorption; and classical and statistical thermodynamics. The chapters that follow focus on the traditional sequence of chemical and phase equilibria, electrochemistry, and chemical kinetics in gas phase and solution phase. This book also considers wave mechanics and its applications; molecular spectroscopy and photochemistry; and the excited state, and then concludes with an analysis of crystal structure, colloid and polymer chemistry, and radio and nuclear chemistry. This reference material is intended primarily as an introductory text for students of physical chemistry.

Solutions Manual For Chemical Engineering Thermodynamics-Y. V. C. Rao 1998 This book is a very useful reference that contains worked-out solutions for all the exercise problems in the book Chemical Engineering Thermodynamics by the same author. Step-by-step solutions to all exercise problems are provided and solutions are explained with detailed and extensive illustrations. It will come in handy for all teachers and users of Chemical Engineering Thermodynamics.

Molecular Physics, Thermodynamics, Atomic and Nuclear Physics-V. L. Ginzburg 2013-10-22 Problems in Undergraduate Physics, Volume IV: Molecular Physics, Thermodynamics, Atomic and Nuclear Physics presents a set of problems in physics as well as answers and solutions in the second part. This book covers several subjects, including thermometry, atoms, kinetic theory of matter, surface tension, thermodynamics, and thermal conductivity. Organized into two parts encompassing two chapters, this volume begins with several problems involving molecular physics, particularly calorimetry, thermal expansion, and thermometry. This text proceeds with a set of problems concerning atomic and nuclear physics, including the quantum nature of light, the wave properties of particles, X-rays, and structure of the atom and spectra. Tables at the end of this book provide information on the range-energy relationships for particles in emulsions as well as well as on the uranium-radium radioactive series. This book is intended to be suitable for students in physics. Teachers and research workers will also find this book extremely useful.

Thermodynamics of Solutions-Eli Ruckenstein 2009-06-17 This book consists of a number of papers regarding the thermodynamics and structure of multicomponent systems that we have published during the last decade. Even though they involve different topics and different systems, they have something in common which can be considered as the "signature" of the present book. First, these papers are concerned with "difficult" or very nonideal systems, i. e. systems with very strong interactions (e. g. , hyd- gen bonding) between components or systems with large differences in the partial molar v- umes of the components (e. g. , the aqueous solutions of proteins), or systems that are far from "normal" conditions (e. g. , critical or near-critical mixtures). Second, the conventional th- modynamic methods are not sufficient for the accurate treatment of these mixtures. Last but not least, these systems are of interest for the pharmaceutical, biomedical, and related ind- tries. In order to meet the thermodynamic challenges involved in these complex mixtures, we employed a variety of traditional methods but also new methods, such as the fluctuation t- ory of Kirkwood and Buff and ab initio quantum mechanical techniques. The Kirkwood-Buff (KB) theory is a rigorous formalism which is free of any of the - proximations usually used in the thermodynamic treatment of multicomponent systems. This theory appears to be very fruitful when applied to the above mentioned "difficult" systems.

Molecular Thermodynamics-James W. Whalen 1991-05-13 Seeking to introduce molecular thermodynamics in a way that is more congruent with the present day, it approaches the subject from a statistical basis, rather than traditional phenomenological bulk phase behavior and continuum mechanics arguments. Thus, topics are discussed in a different sequence than is encountered in more traditional texts; the presentation of material begins with the molecular argument and later expands to bulk phase behavior. Chapters cover thermal and mechanical processes, structured particle systems and interacting particle systems, multicomponent systems, macroscopic process considerations, electrolyte systems, and more. Worked examples and end-of-chapter problems are included.

Thermodynamics and Statistical Mechanics-M. Scott Shell 2015-04-16 Learn classical thermodynamics alongside statistical mechanics and how macroscopic and microscopic ideas interweave with this fresh approach to the subjects.

Physical Chemistry-Andrew Cooksy 2014 In the phase transitions among the solid, liquid, and gaseous forms of water, we see a profound demonstration of how properties at the molecular scale dictate the behavior of the bulk material. As ice is heated beyond its melting point, new avenues for molecular motion become open to the energy being added. Upon entering the gas phase, the water molecules can explore new territory, unavailable to the liquid or solid. These transformations can be seen as a shifting balance between the forces that bind the molecules and the thermal energy that excites these motions--a window through thermodynamics on the intricate mechanisms that drive chemistry.

Physical Chemistry of Polymer Solutions-K. Kamide 2000-10-16 This book is mainly concerned with building a narrow but secure ladder which polymer chemists or engineers can climb from the primary level to an advanced level without great difficulty (but by no means easily, either). This book describes some fundamentally important topics, carefully chosen, covering subjects from thermodynamics to molecular weight and its distribution effects. For help in self-education the book adopts a "Questions and Answers" format. The mathematical derivation of each equation is shown in detail. For further reading, some original references are also given. Numerous physical properties of polymer solutions are known to be significantly different from those of low molecular weight solutions. The most probable explanation of this obvious discrepancy is the large molar volume ratio of solute to solvent together with the large number of consecutive segments that constitute each single molecule of the polymer chains present as solute. Thorough understanding of the physical chemistry of polymer solutions requires some prior mathematical background in its students. In the original literature, detailed mathematical derivations of the equations are universally omitted for the sake of space-saving and simplicity. In textbooks of polymer science only extremely rough schemes of the theories and then the final equations are shown. As a consequence, the student cannot learn, unaided, the details of the theory in which he or she is interested from the existing textbooks; however, without a full understanding of the theory, one cannot analyze actual experimental data to obtain more basic and realistic physical quantities. In particular, if one intends to apply the theories in industry, accurate understanding and ability to modify the theory are essential.

Phase Diagrams and Thermodynamic Modeling of Solutions-Arthur D. Pelton 2018-09-19 Phase Diagrams and Thermodynamic Modeling of Solutions provides readers with an understanding of thermodynamics and phase equilibria that is required to make full and efficient use of these tools. The book systematically discusses phase diagrams of all types, the thermodynamics behind them, their calculations from thermodynamic databases, and the structural models of solutions used in the development of these databases. Featuring examples from a wide range of systems including metals, salts, ceramics, refractories, and concentrated aqueous solutions, Phase Diagrams and Thermodynamic Modeling of Solutions is a vital resource for researchers and developers in materials science, metallurgy, combustion and energy, corrosion engineering, environmental engineering, geology, glass technology, nuclear engineering, and other fields of inorganic chemical and materials science and engineering. Additionally, experts involved in developing thermodynamic databases will find a comprehensive reference text of current solution models. Presents a rigorous and complete development of thermodynamics for readers who already have a basic understanding of chemical thermodynamics Provides an in-depth understanding of phase equilibria Includes information that can be used as a text for graduate courses on thermodynamics and phase diagrams, or on solution modeling Covers several types of phase diagrams (paraequilibrium, solidus projections, first-melting projections, Scheil diagrams, enthalpy diagrams), and more

Molecular Driving Forces-Ken Dill 2010-10-21 Molecular Driving Forces, Second Edition E-book is an introductory statistical thermodynamics text that describes the principles and forces that drive chemical and biological processes. It demonstrates how the complex behaviors of molecules can result from a few simple physical processes, and how simple models provide surprisingly accurate insights into the workings of the molecular world. Widely adopted in its First Edition, Molecular Driving Forces is regarded by teachers and students as an accessible textbook that illuminates underlying principles and concepts. The Second Edition includes two brand new chapters: (1) "Microscopic Dynamics" introduces single molecule experiments; and (2) "Molecular Machines" considers how nanoscale machines and engines work. "The Logic of Thermodynamics" has been expanded to its own chapter and now covers heat, work, processes, pathways, and cycles. New practical applications, examples, and end-of-chapter questions are integrated throughout the revised and updated text, exploring topics in biology, environmental and energy science, and nanotechnology. Written in a clear and reader-friendly style, the book provides an excellent introduction to the subject for novices while remaining a valuable resource for experts.

Thermodynamics of Polymer Solutions-Michio Kurata 1982

Molecular Thermodynamics of Aqueous Systems Containing Hydrocarbons, Polymers, Or Gels-Herbert Hearl Hooper 1990

Molecular Thermodynamics of Liquid-liquid Equilibria in Polymer Mixtures-Stephen Mark Lambert 1995

Molecular Thermodynamics-John H. Knox 1971

Classical Thermodynamics of Non-Electrolyte Solutions-H. C. Van Ness 2015-12-04 Classical Thermodynamics of Non-Electrolyte Solutions covers the historical development of classical thermodynamics that concerns the properties of vapor and liquid solutions of non-electrolytes. Classical thermodynamics is a network of equations, developed through the formal logic of mathematics from a very few fundamental postulates and leading to a great variety of useful deductions. This book is composed of seven chapters and begins with discussions on the fundamentals of thermodynamics and the thermodynamic properties of fluids. The succeeding chapter presents the equations of state for the calculation of the thermodynamic behavior of constant-composition fluids, both liquid and gaseous. These topics are followed by surveys of the mixing of pure materials to form a solution under conditions of constant temperature and pressure. The discussion then shifts to general equations for calculation of partial molal properties of homogeneous binary systems. The last chapter considers the approach to equilibrium of systems within which composition changes are brought about either by mass transfer between phases or by chemical reaction within a phase, or by both.

Solutions and Solubilities-Michael R. J. Dack 1975

The Molecular Theory of Solutions-Ilya Prigogine 1957

Engineering and Chemical Thermodynamics-Milo D. Koretsky 2012-12-17 Chemical engineers face the challenge of learning the difficult concept and application of entropy and the 2nd Law of Thermodynamics. By following a visual approach and offering qualitative discussions of the role of molecular interactions, Koretsky helps them understand and visualize thermodynamics. Highlighted examples show how the material is applied in the real world. Expanded coverage includes biological content and examples, the Equation of State approach for both liquid and vapor phases in VLE, and the practical side of the 2nd Law. Engineers will then be able to use this resource as the basis for more advanced concepts.

Basic Chemical Thermodynamics-Eric Brian Smith 2004 This widely acclaimed text, now in its fifth edition and translated into many languages, continues to present a clear, simple and concise introduction to chemical thermodynamics. An examination of equilibrium in the everyday world of mechanical objects provides the starting point for an accessible account of the factors that determine equilibrium in chemical systems. This straightforward approach leads students to a thorough understanding of the basic principles of thermodynamics, which are then applied to a wide range of physico-chemical systems. The book also discusses the problems of non-ideal solutions and the concept of activity, and provides an introduction to the molecular basis of thermodynamics.Over five editions, the views of teachers of the subject and their students have been incorporated. The result is a little more rigour in specifying the dimensions within logarithmic expressions, the addition of more worked examples and the inclusion of a simple treatment of the molecular basis of thermodynamics. Students on courses in thermodynamics will continue to find this popular book an excellent introductory text.

Physical Chemistry: A Molecular Approach-Donald A. McQuarrie 1997-08-20 As the first modern physical chemistry textbook to cover quantum mechanics before thermodynamics and kinetics, this book provides a contemporary approach to the study of physical chemistry. By beginning with quantum chemistry, students will learn the fundamental principles upon which all modern physical chemistry is built. The text includes a special set of "MathChapters" to review and summarize the mathematical tools required to master the material Thermodynamics is simultaneously taught from a bulk and microscopic viewpoint that enables the student to understand how bulk properties of materials are related to the properties of individual constituent molecules. This new text includes a variety of modern research topics in physical chemistry as well as hundreds of worked problems and examples.

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